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(54) Title: CYCLIC ADENOSINE DIPHOSPHATE RIBOSE ANALOGUES

(57) Abstract

A compound having formula (1) where at least one of X^3 and X^7 is CR and any remaining X^3 or X^7 is N; Y is halo, C1–C20 hydrocarbon, NR₂, OR, SR, nitro or carboxyl; R is H or C1–C20 hydrocarbon and R groups can be the same or different; and one or each Z represents H or one Z is a cageing group. The compound 7–deaza–8–bromo–cyclic adenosine 5'-diphosphate ribose has been shown to be a stable hydrolysis–resistant cADPR antagonist.

$$ZO-P-O$$
 X_3
 X_7
 $ZO-P-O$
 Y_7
 Y_7

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CYCLIC ADENOSINE DIPHOSPHATE RIBOSE ANALOGUES

INTRODUCTION

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Cyclic adenosine diphosphate ribose (cADPR) is a ubiquitous Ca²⁺ mobilizing metabolite of β-NAD⁺ (1, 2). It is reported to mediate Ca²⁺ release via ryanodine sensitive channels in many cell types, in both animal and plant kingdoms (1, 3-6). Endogenous levels of cADPR have been detected and reported to be equally widespread (7). This finding has led to the postulation that cADPR may be the endogenous/physiological regulator of ryanodine receptors (5, 8-10).

As is the case for the more established intracellular messengers (i.e. IP₃ cAMP and cGMP), cADPR metabolising enzymes are also present which can modulate its levels (11). The synthetic activity of ADP-ribosyl cyclase and catabolic activity of cADPR hydrolase are often co-localised on the same polypeptide. In these cases, the hydrolase activity often exceeds that of cyclase (1). However, one notable exception is *Aplysia* ADP-ribosyl cyclase which is isolated and purified from soluble ovotestis extracts of the sea hare. *Aplysia californica*. The exceptionally high level of cyclase activity exhibited by this enzyme (1) has been well exploited to synthesise large quantities of cADPR. In addition, the finding that this cyclase exhibits loose substrate specificity, has allowed the development of a chemo-enzymatic synthesis of a number of cADPR analogues (12).

The first series of pharmacologically useful cADPR analogues to be synthesised were the 8-substituted analogues (13). These differ from cADPR by a substitution at the 8 position of the adenine ring. This single modification abolishes the agonistic activity of these compounds and instead produces specific competitive antagonists of

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cADPR-sensitive Ca²⁺ release (13). Since its discovery, 8-NH₂-cADPR has been used successfully to demonstrate the involvement of cADPR - mediated Ca²⁺ signalling in sea urchin eggs during fertilisation (14) and NO and cGMP-induced Ca²⁺ release (15), in Purkinje neurons (16), in hippocampal synaptic plasticity, in permeabilized Jurkat T cells (6), in intestinal smooth muscle during cholecystokinin-induced contractions (5), in PC12 cells (17) and in excitation contraction coupling in cardiac myocytes (18). However, like the parent compound, cADPR, the 8-substituted analogues are prone to hydrolysis by endogenous enzymes (13). Indeed, this may explain the absence of inhibitory effect on secretogue-induced Ca²⁺ release in rat pancreatic beta-cells (19) and during induction of LTD in Purkinje neurons (16), where a role for cADPR-mediated Ca²⁺ signalling remains controversial.

These observations underscore the need for a stable, hydrolysis-resistant cADPR antagonist. Recently, we reported on the synthesis of another analogue of cADPR, 7-deaza-cADPR, and demonstrated that it is more stable to heat-induced hydrolysis and is also a poor substrate for cADPR hydrolase (20). These changes in stability were also brought about by a single modification, that is a replacement of the 7 position nitrogen with carbon (Figure 1A). These findings then raised an intriguing question: what would be the biological activity of a compound modified at both the 7 and 8 positions of the adenosine ring? A hybrid' analogue, was successfully synthesised namely, 7-deaza-8-bromo-cADPR (Figure 1A). Its biological properties were examined and are reported herein. Our findings show that 7-deaza-8-bromo-cADPR retains both useful pharmacological properties i.e. it is a hydrolysis resistant, antagonist of cADPR-induced Ca2+ release. Furthermore, owing to the lipophilic nature of both the bromo- and CH- moieties, we have explored its potential as a membrane permeable analogue of cADPR, as has been established for 8-bromo-cGMP (Br-cGMP) c.f. cGMP (21). A single molecular species

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exhibiting all three properties would be a very powerful pharmacological tool for investigations of cADPR-mediated Ca²⁺ signalling in intact cells. We report here that 7-deaza-8-bromo-cADPR is such a tool.

5 ABBREVIATIONS:

cADPR cyclic adenosine 5'-diphosphate ribose;

ADPR Adenosine 5'-diphosphate ribose;

IP₃ myo-inositol-(1,4,5)-trisphosphate;

cAMP cyclic adenosine 3', 5'-monophosphate;

cGMP cyclic guanosine 3', 5'-monophosphate;

CICR calcium-induced calcium release

ASW artificial sea water

IM intracellular medium.

The invention provides compounds having the formula (1)

where at least one of X3 and X7 is CR and any remaining X3

20 or X^7 is N,

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Y is halo, C1 - C20 hydrocarbon, NR₂, OR, SR, nitro or

carboxyl,

R is H or C1 - C20 hydrocarbon and R groups can be the

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same or different,

and one or each Z represents H or one Z is a cageing group.

Preferably X⁷ is CH and X³ is N. The preparation of such a compound is described below. Preparation of compounds where X⁷ is not CH, or where X³ is not N can readily be achieved by known chemistry using a different starting compound 6 (see the reaction scheme below).

Y is preferably halo, particularly bromo. This is the compound whose preparation is described in detail below. Alternatively Y may be amino. Alternatively Y may be iodo, either a natural or a radioisotope such as ¹²⁵I. Radiolabelled compound may be useful in an assay, eg. a competitive binding assay, to screen for other compounds with more potent binding characteristics for a cADPR receptor. Compounds where Y is other than bromo may readily be prepared *inter alia* using known chemistry with a different n-succinimide derivative in reaction (i).

R is H or C1 - C20 hydrocarbon e.g. alkyl, aryl, aralkyl, alkaryl, alkenyl or alkynyl and which may be substituted or unsubstituted. Preferably R is H or C1 - C4 alkyl.

Either both groups Z represent hydrogen atoms (or negative charges) as in a conventional diphosphate system. Or one group Z may be a "cageing" group e.g. 1-(o-nitrophenyl)ethyl, whereby the compound of the invention is a caged analogue adapted to be converted into an active compound *in situ* e.g. by light irradiation. The chemistry of such caged analogues is well known, see Aarhaus R *et al*, J. Biol. Chem., 1995, 270, 7745; McCray J A and Trentham D R, Ann. Rev. Biophys., Biophys. Chem, 1989, 18, 239-270 and such compounds can be prepared by standard methods.

Although the two sugar rings have been shown as ribose for simplicity, it is envisaged that other groups may be present as in known nucleotide chemistry. Thus, on either ribose, the 5'- oxygen atom may be

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replaced by S, CHF, CF₂, SO, NR or CH₂. Also the 2'- and 3'-hydroxyl groups may be replaced by H, F, NH₂, N₃ or O-hydrocarbyl. Such compounds may be made by known chemical methods analogous to those described below.

MATERIALS AND METHODS

Chemo-enzymatic synthesis of 7-deaza-8-bromo-cADPR.

The synthesis of 7-deaza-8-bromo-cADPR. Briefly, 7-deaza-adenosine (Tubercidin) was brominated and selectively phosphorylated to yield 7-deaza-8-bromo-adenosine 5'-monophosphate (7-deaza-8-bromo-AMP). This was then coupled to nicotinamide mononucleotide (NMN) to form 7-deaza-8-bromo-β-NAD⁺ as has previously been described for 7-deaza-β-NAD⁺(20). The enzymatic cyclization of 7-deaza-8-bromo-NAD⁺ to 7-deaza-8-bromo-cADPR was catalysed by crude *Aplysia* ADP-ribosyl cyclase and the product was purified by ion-exchange chromatography (12, 20). The extinction coefficient for 7-deaza-8-bromo-cADPR was determined by total phosphate analysis to be 10.85 x 10³ M⁻¹cm⁻¹ (277 nm) and this value was used to determine the concentrations of 7-deaza-8-bromo-cADPR used in the rest of the study.

The reaction scheme below shows synthesis of 7-deaza-8-bromo-cADPR 5. Reagents and Conditions: (i) N-bromosuccinimide, K* acetate, DMF, 10 min, 40%; (ii) (a) PO(Oet)₃ 50°C, 15 min; (b) POCl₃ 0°C, 2h; (c) H₂O, 44%; (iii) DCC, pyridine:H₂O 4:1, 7d, 28%; (iv) 1.5mM solution in 25 mM 2-[4-(2-hydroxyethyl)piperazin-1-yl] ethanesulfonic acid, ADP-ribosyl cyclase, 30 min, 31%.

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Preparation of 7-Deza-8-Bromo-Adenosine (7)

To a stirred solution of dry 7 deazaadenosine (Tubercidin, 235mgs, 0.88mmol) and dry potassium acetate (185mgs, 1.88 mmol) in DMF (3.8ml) was added dropwise a solution of freshly recrystallised

N-bromosuccinimide (335mgs, 1.88mmol) in DMF (1.9ml). The reaction mixture immediately turned dark red and was stirred for a further 10 min. The DMF was removed by evaporation *in vacuo* and the residue dissolved in methanol and adsorbed onto silica gel. Flash chromatography was used to purify the product with elution, using a gradient of 0-10% MeOH/CHCl₃.

Combination of the fractions containing material with a R_f of 0.07 (10%MeOH/CHCl₃) yielded a yellow solid product (57) in 40% yield. This was used in the next step without further purification.

NMR δ_H (400 MHz, d_6 DMS0, ppm); 3.48 (1H. m H5'; D_2 O ex, ABX, J_{AB} =12.2, J_{AX} =3 9Hz) 3.66 (1H, m, H5'; D_2 O ex, ABX, J_{AB} =12.2, J_{AX} =3.9Hz), 3-92 (1H, dt, J=3.9, 2.4Hz, H4'), 4.14 (1H, m, H3'; D_2 O ex, dd, J=5.7, 2.4Hz), 5.13 (2H, m, H2' and 3'OH; D_2 O ex, dd, J=6.8, 5.7Hz, 1H), 5.31 (1H, d, J=6.6Hz, D_2 O ex, 2'0H), 5.70 (1H, dd, J=5.6, 3.8Hz D_2 O ex, 5'OH), 5.84 (1H, d, J=6.8Hz) H1'), 6.80 (1H. s, H7), 7.29 (2H, br s. D_2 O ex, NH₂), 8.02 (1H, s, H2).

NMR δ_{c} (100.4 MHz, d_{6} DMSO, ppm); 62.7 (C5'), 71.3 (C3'), 71.6 (C2'), 86.5 (C4'), 90.4 (C1'), 103.3 (C7), 104.3 (C5), 109.7 (C8), 149.9 (C4), 151.9 (C2), 156.9 (C6),

m/z (FAB⁺) 345, 347 (M⁺+1), 213, 215 (M⁺+1-ribose).

25 Preparation of 7-Deaza-8-Bromo~Adenosine 5'-Monophosphate (8)

To a cooled solution of dry 7-deaza-8-bromo-adenosine (100 mg 290 μ mol) suspended in triethyl phosphate (2ml) was added POCl₃ (100 μ l, 500 μ mol) and the reaction was monitored by HPLC (starting material 1.74min, product 2.15min). After 90min cooled 1:9 pyridine: water (10ml) was added to the above followed by stirring for a

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further 30min. This product was purified by ion exchange chromatography and the fractions containing the product, which eluted between 310 and 370mM TEAB, were pooled evaporated *in vacuo* to give the pure product in 44% yield.

NMR δ_H (400 MHz, D₂O) 4.08-4.15 (3H. m, H4' and H5'), 4.53 (1H, dd, J=5.7, 5.8Hz, H3'), 5 07 (1H, dd, J=5.3, 5.8Hz, H2'), 6.04 (1H, d, J=5.3Hz, H1'), 6.42 (1H, s, H7), 7.92 (1H, s, H2),

 δ_{P} (161.7MHz, $D_{2}O);$ 2.35ppm, broadens when proton coupled,

 δ_{C} (100.4MHz, D₂O); 65.2 (d. J_{CP}=3.7Hz, C5'), 70.3 (C3'), 72.1 (C2'), 83.6 (d, J_{CP}=7.4Hz, C4'), 89.8 (C1'), 104.0 (C5), 105.2 (C7), 110.4 (C8), 149.9 (C4). 150.0 (C2), 154.5 (C6),

UV: λ_{max} 276 nm, ϵ =13.34x10³ -1 cm⁻¹.

HPLC retention time: product 2.151min, m/z (FAB⁺) 425, 427 (M⁺+I); (FAB⁻) 423, 425 (M⁺-2), 341 (M⁺-Br-3).

Preparation of 7-Deaza-8-Bromo-Nicotinamide Adenine Dinucleotide (7-Deaza-8-Bromo-NAD) (9)

To a solution of 7-deaza-8-bromo-adenosine 5'-monophosphate (94μmol) and NMN (20mg, 60μmol) in MilliQ water (1ml) was added pyridine (4ml) and an excess of DCC (0.5g) and the reaction was stirred at room temperature for 7 days. The reaction was quenched by the addition of MilliQ water (50ml), the mixture kept at 4°C for 2h, filtered and extracted with CHCl₃ (3x20ml). The material was purified by ion exchange chromatography and the fractions containing the product, which eluted between 90 and 120mM TEAB, were combined, the solvent removed *in vacuo* and excess TEAB removed by coevaporation with MeOH to yield the title compound in 27.7% yield.

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 $(M^{+}-1)$.

NMR: δ_H (400MHz, D_2O); 4.03-4.32 (8H, m, H_N2' , H_N3' , H_A4' , H_N4' , H_N4' , H_N4' , H_N4' , H_N4' , H_N4' , H_N5' ,

 $\delta_{P} \ (161\ 7\text{MHZ},\ D_{2}\text{O});\ -10.5\ -10.9\text{ppm}\ (2d,\ J_{PP}=20.1\text{HZ}),$ $\delta_{C} \ (100.4\ \text{Hz},\ D_{2}\text{O});\ 65\ 7\ (d,\ J=3.7\text{Hz},\ C_{A}5'),\ 66.5\ (C_{N}5')\ 70.0$ $(C_{A}3')\ 71\ 4\ (C_{N}3'),\ 7\ 1.9\ (C_{A}2'\),\ 78.5\ (C_{N}2'\),\ 83.5\ (d,\ J=9.2\text{Hz},\ C_{A}4'),\ 87.8$ $(d,\ J=9.2\text{Hz},\ C_{N}4'\),\ 89.8\ (C_{A}1'),\ 100.9\ (C_{N}1'),\ 104.3\ (C_{A}5),\ 105.0\ (C_{A}7),$ $110.5\ (C_{A}8),\ 129\ 6\ (C_{N}4),\ 134.2\ (C_{N}3),\ 140.6\ (C_{N}5),\ 143.1\ (C_{N}6),\ 146.3$ $(C_{N}2),\ 150.4\ (C_{A}2),\ 150.7\ (C_{A}4),\ 155.5\ (C_{A}6),\ 166.1\ (CO),$

UV: λ_{max} =270nm, ϵ =13.3x10³M⁻¹cm⁻¹,

HPLC retention time 2.1 6min,

m/z (electrospray⁺) 741, 743 (M⁺+I); (electrospray⁻) 739, 741

Preparation of 7-Deaza-8-Bromo Cyclic Adenosine Diphosphate Ribose (7-Deaza-8-bromo-cADPR) (5)

7-Deaza-8-bromo NAD* (triethylammonium salt, 12μmol)) was cyclised as described and the product 7-deaza-8-bromo-cADPR which eluted between 130 and I150mM TEAB, was evaporated to dryness *in vacuo* keeping the water bath temperature below 30°C. Excess TEAB was removed by evaporation with MeOH and the cyclic compound was isolated as the glassy triethylammonium salt in 31% yield as quantified by UV and stored at -70°C.

NMR $\delta_{\rm H}$ (400MHz, D₂O); 3.89-3.93 (1H, m, H_A5'), 3.98-4.03 (1H, m, H_R5'), 4 1-4.20 (1H, m, H_A4'),4.23-4.27 (1H, m, H_R5'), 4.30-4.40 (2H, m, H_A5', H_R3'), 5.44 (1H, dd, J=5.2, 6.1Hz, H_A2'), 5.97 (1H, d, J=4.0Hz, H_R1'), 6.03 (1H, d, J=6.1Hz, H_A1'). 6 91 (1H, s, H_A7), 8 75 (1 H, s, H_A2), all other protons are obscured by the water peak at 4 8ppm

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 δ_{P} (161. 7MHz, D₂O); -10.8, -11.6ppm (2d, J_{PP}=18HZ), UV: λ_{max} =277Dm, ϵ =10-85x10³M⁻¹Cm⁻¹, HPLC retention time 4.8min, m/z (electrospray⁺) 619, 621 (M⁺+1); (electrospray⁻) 617, 619 (M⁺-1), 79, 81 (Br⁻).

In vitro Ca2+ release assays

In vitro Ca²⁺ release assays were performed on sea urchin egg homogenates (2.5%, v/v) prepared from unfertilised *Lytechinus pictus* eggs according to the method of Clapper and Lee (22) with modifications as previously described (23). Extramicrosomal Ca²⁺ was thus measured by monitoring changes in fluo-3 (3 μ m) fluorescence (excitation 490 nm and emission 530 nm) in a Perkin-Elmer LS-SOB fluorimeter. All additions (not exceeding 5 μ l) were made to cuvettes (containing 500 μ l homogenate) in intracellular medium (1M) (containing 10 μ m EGTA).

Intracellular free Ca2+ measurements in intact eggs

 Ca^{2^+} imaging of intact cells was performed using unfertilised Lytechinus pictus eggs micro-injected with 2 μM fura-2 and 250 $\mu\text{g/ml}$ heparin as previously described (23, 24).

Materials

Lytechinus pictus sea urchins were from Marinus Inc. Long Beach, CA. USA. Fluo-3 and Fura 2 was purchased from Cambridge Bioscience (Molecular Probes). All other chemicals were from Sigma Inc. (London). Cyclic ADP-ribose, 8-bromo-cADPR and 7 deaza-cADPR were synthesised as previously described (12, 20).

RESULTS & DISCUSSION

Since the elucidation of the structure of cADPR (25, 26) and

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the development of a chemoenzymatic synthesis for cADPR and its analogues (12), interest has mounted concerning the structure-function relationships between the ligand and its endogenous receptor(s). To further aid these diagnostic studies, we synthesised 7-deaza-8-bromocADPR. a novel analogue of cADPR. Figure 1A shows the chemical structure of 7-deaza-8-bromo-cADPR as compared with that of cADPR and two previously reported analogues 8-bromo-cADPR, a specific cADPR antagonist (10) and 7-deaza cADPR, a hydrolysis resistant, partial agonist (20). Whereas the latter two compounds have a single modification made at either the 7 or 8 position of the adenosine ring, 7-deaza-8-bromocADPR possesses both these modifications. The effect of each modification on the Ca2+ mobilizing (agonistic) ability was first tested using sea urchin egg homogenates (2.5%). Figure 1B shows the Ca2+ releasing. action of 2 µM applications of 7-deaza-8-bromo-cADPR and related cyclic nucleotides. Unlike the agonists cADPR and 7-deaza cADPR, 7-deaza-8bromo-cADPR did not induce Ca2+ release from sea urchin egg microsomes, even at concentrations up to 20µM (a supra-maximal agonist concentration). In this respect 7-deaza-8-bromocADPR resembles the antagonist 8-bromo-cADPR.

Whether 7-deaza-8-bromo-cADPR, like 8-bromo-cADPR, was also an antagonist of cADPR-sensitive Ca²⁺ release was investigated next. Figure 2 shows that this was indeed the case. Sea urchin egg homogenates pre-treated with either 8-bromo-cADPR or 7-deaza-8-bromo-cADPR were markedly less responsive to 100 nM cADPR (Figure 2A).

These inhibitory actions were dependent on the antagonist concentration (Figure 2B). Both 8-bromo-cADPR and 7-deaza-8-bromo-cADPR exhibited similar inhibition potencies with comparable IC₅₀ values (IC₅₀ = $0.97 \pm 0.04 \mu m$ (S.E. n=3) for 8-bromo-cADPR: IC₅₀= $0.73 \pm 0.05 \mu M$ (S.E. n=3) for 7-deaza-8-bromo-cADPR.).

Lower concentrations (31 mM) of either antagonist were also

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significant in preventing Ca²⁺ release by 100 nM cADPR which did not exceed 85% of control values. Whether this is due to the presence of more than one population of receptors which exhibit different binding affinities and/or sensitising properties remains to be investigated.

Nonetheless, both 8-substituted analogues appeared to behave similarly with respect to these actions on the cADPR-induced Ca²⁺ release channel in sea urchin egg homogenates. As previous studies have shown that 8-substituted cADPR analogue (7,13, [new 14] and 7-deaza-cADPR (20) are able to displace cADPR binding, it is likely that 7-deaza-8-bromo-cADPR also interacts at the cADPR receptor in the same specific manner.

Since a modification on the 8-position does not alter the stability of the molecule (13) but a substitution of N⁷ with a carbon has been shown to render the cyclic compound more resistant to hydrolysis (20), we investigated whether, 7-deaza-8-bromo-cADPR could differ from 8-bromocADPR, but resemble 7-deaza-cADPR in this respect. We subjected standard solutions of both antagonists to heat-induced hydrolysis. This treatment has previously been shown to strip unstable cyclic compounds such as cADPR and 8 NH2-cADPR, of their biological activity (13, 18, 20). Figure 3 shows the effect of heat treatment on the antagonistic actions of both 7-deaza-8-bromo-cADPR and 8-bromo-cADPR. Whereas 8-bromocADPR is stripped of its antagonistic activity, 7-deaza-8-bromo-cADPR remained an effective antagonist of cADPR-induced Ca2+ release (Figure 3B compared to A). HPLC analysis of the same samples confirmed that this loss of activity was due to degradation of 8-bromo-cADPR whilst 7-deaza-8-bromo-cADPR remained intact (data not shown). These data are in line with the chemical stability of 7-deaza adenosine c.f. adenosine (27) and suggests that 7-deaza-8-bromo-cADPR is the first stable cADPR antagonist. This new compound therefore has potential for application in intact cells/tissues such as Jurkat T cells and neuronal tissue which express high hydrolase activities but may still have a functional cADPR-

mediated Ca2+ signalling pathway (1).

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When these antagonists were tested for resistance to enzyme mediated hydrolysis (by cADPR hydrolases) a similar resistance emerged. Egg homogenates (2.5%. containing fluo-3), were incubated overnight (at 17°C) with 20 μM 8-bromo-cADPR or 7-deaza-8-bromo-cADPR. Samples (50 μI) were then taken and tested for antagonistic activity on cADPR-induced Ca²+ release (as described for figure 2 but where the cuvette concentration of the antagonists was initially 2 μM). Whereas the antagonistic activity of 8-bromo-cADPR, in response to cADPR (100 nM), had dramatically reduced following the prolonged incubation with *L. pictus* cADPR hydrolase (65.8 ± 8.3% of control cADPR-induced Ca²+ release, n=3), the levels of 7-deaza-8-bromo-cADPR remained high thereby producing a greater antagonistic effect on cADPR-induced Ca²+ release (28.5 ± 11.4% of control cADPR-induced Ca²+ release (28.5 ± 11.4% of control

It has been demonstrated that the presence of a lipophilic bromine moiety in cGMP affords greater membrane permeability to BrcGMP (21) and the replacement of a nitrogen with CH- group also offers greater hydrophobicity (28). Therefore, the novel cADPR analogue, 7-deaza-8-bromo-cADPR, should have greater hydrophobic character than any previously synthesised. We investigated this by testing the effect of extracellular applications of 7-deaza-8-bromo-cADPR on fertilisation-induced Ca²+ mobilisation in intact sea urchin eggs. Eggs were co-injected with the 1P3 receptor antagonist, heparin (250 μ g/ml) and Ca²+ sensitive fluorochrome fura-2 (2 μ M). Figure 4A shows that upon sperm addition to control heparinized eggs, a propagating Ca²+ wave was produced (Figure 4A, open squares). This Ca²+ wave had an average amplitude of 1705 \pm 119 nM Ca²+ (S.E.M, n = 11) and took 41.9 \pm 5.8 s (S.E.M, n = 11) to reach this peak. These data are consistent with previously reported

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observations of sperm-induced Ca2+ mobilization from IP3-insensitive Ca2+ stores (14, 29). When heparinized eggs were pre-incubated (5 minutes) with bathing solution containing 7-deaza-8-bromo-cADPR (100 μM), the sperm-induced intracellular Ca2+ transients were completely abolished (figure 4A, filled circles) suggesting inhibition of the redundant cADPRsensitive Ca2+ release mechanism (14., 29). In contrast, when 8-bromocADPR was present in the bathing medium, a Ca2+ transient was still observed, albeit dramatically reduced (figure 4A, filled triangles). The wave property of the Ca2+ response was also abolished by both treatments where the Ca2+ rises occurred slowly (if at all) and over the whole egg. Although the inhibition of fertilisation-induced Ca2+ transients was dependent on antagonist concentration (figure 4B. filled symbols), 7-deaza-8-bromo-cADPR appeared to be the more effective antagonist. Neither antagonist released Ca²⁺ in the eggs during the 5 minute pre-incubation period which is consistent with the absence of agonistic activity observed In vitro (figure 1B).

Since intracellular Ca²⁺ mobilization is a pre-requisite for the cortical reaction (30, 31), we also monitored the formation of activation envelopes following fecundation. Treatment of eggs with either 7-deaza-8-bromo-cADPR or 8-bromo-cADPR in the bathing medium, was found to prevent the cortical reaction (in heparinized eggs) in a concentration dependent manner (Figure 4B, open symbols). This is in keeping with previous reports which demonstrate inhibition of the cortical reaction only once both redundant mechanisms have been blocked (14, 29). Eggs that were treated with antagonist only (i.e. not micro-injected with heparin), were also scored for activation envelopes in the same experiments. These consistently showed >95% activation. This suggested that neither 7-deaza-8-bromo-cADPR nor 8-bromo-cADPR acted as spermicides, rather they are able to permeate the sea urchin egg plasma membrane and specifically compete for endogenous cADPR-binding sites and inhibit

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agonist-induced Ca²⁺ mobilisation. Presumably, since the net charge on 7-deaza-8-bromo-cADPR is only -1 at physiological pH (figure 1A), this compares well with 8-bromo-cGMP which has the same net charge and where 8-substitution confers membrane permeability. This property now eliminates the need for potentially disruptive protocols such cell permeabilization or micro-injection as methods for introducing cADPR antagonists into whole cells. This advancement should greatly aid investigations of the role of c.ADPR in physiological responses to extracellular stimuli.

In conclusion, we have demonstrated that, unlike 8-bromo-cADPR, 7-deaza-8-bromo-cADPR is a stable hydrolysis-resistant cADPR antagonist. This is the first report of such a compound. In addition, we have exploited the lipophilic nature of the bromo-moiety to produce a compound that is also sufficiently membrane permeable. In all this makes 7-deaza-8-bromo-cADPR a very powerful pharmacological tool for investigations of cADPR-mediated Ca²⁺ signalling in intact cells.

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FIGURE LEGENDS

- Figure 1: (A) Structural formulae of cADPR 7-deaza-cADPR, 8-bromo-cADPR and 7-deaza-8-bromo-cADPR. Modifications were made to cADPR at the 7 and 8 positions of the adenosine ring. The N⁷ nitrogen atom is replaced with a carbon (and associated proton) to form 7-deaza
 - cADPR whilst 8-bromo-cADPR is formed by substituting the hydrogen on the C-8 position with a bromine atom. 7-Deaza-8-bromo-cADPR has both
- these modifications. (B) Calcium releasing action of cADPR, 7-deaza-

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homogenates. *Lytechinus pictus* egg homogenates (2.5%, v/v) containing the Ca²⁺ sensitive dye. fluo-3 (3 μ M) was prepared as described in Materials and Methods. Aliquots (500 μ I) of these were challenged with 5 μ I of cyclic compounds to give a final cuvette concentration of 2 μ M. The Ca²⁺ release profiles observed are shown and are representative of three separate experiments.

Figure 2: Concentration dependent inhibition of cADPR-induced Ca²⁺ release by 8-bromo-cADPR and 7-deaza-8-bromo-cADPR.

Lytechinus pictus egg homogenates (2.5 %, v/v) containing the Ca²+ sensitive dye, fluo-3 (3 μ M) were prepared as described in Materials and Methods. Their sensitivity to cADPR-induced Ca²+ release was used to test the antagonistic action of 8-bromo-cADPR (A) as compared to that of 7-deaza-8-bromo-cADPR (B). This was done by pretreating homogenates with increasing concentrations of antagonist (in 5 μ l Glu IM + EGTA), 3 minutes prior to challenge with 100 nM cADPR. Representative traces are shown in (A) and (B) where the addition artefact has been removed for clarity (gap in traces). The amount of Ca²+ released by cADPR application was determined and expressed as a percentage of control cADPR release, i.e. the amount of Ca²+ released in homogenates treated with vehicle. The inhibition curves that were obtained for each antagonist are shown in (C). Each value represents the mean \pm SE of triplicates. The IC₅₀ values were estimated to 0.97 \pm 0.04 μ M for 8-bromo-cADPR and 0.73 \pm 0.05 μ M for 7-deaza-8-bromo-cADPR (S.E. n=3).

Figure 3: Effect of heat treatment on antagonist activity of 8-bromocADPR and 7-deaza-8-bromocADPR.

Experimental conditions were as for figure 2. Complete inhibition of cADPR-induced release was seen when homogenates were pre-treated with 10 μ M of either 8-bromo-cADPR or 7-deaza-8-bromo-

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cADPR (A). The stock solutions (1 mM) of either 8-bromo-cADPR or 7-deaza-8-bromo-cADPR were also incubated at 85°C for 90 min in a water bath. After heat-induced hydrolysis, the antagonistic action of 8-bromo-cADPR was abolished (B). In contrast, heat-treated 7-deaza-8-bromo-cADPR remained an effective antagonist to cADPR-induced Ca²⁺ release. Fluorimetric traces are representative of three similar experiments.

Figure 4. Antagonistic actions of extracellularly applied 7-deaza-8-bromo-cADPR and 8-bromo-cADPR on fertilisation-induced Ca²⁺ transients in intact sea urchin eggs.

(A). Control response (open squares) showing a typical fertilisation-induced Ca2+ transient in intact L. pictus eggs pre-injected with heparin and fura-2 (final concentrations in egg were approximately 250 μ g/ml and 2 μ M respectively). Sperm were added at time = 0. Despite the presence of heparin, an IP3 receptor antagonist, the wave properties of the sperm-induced Ca2+ rise were intact and activation envelopes were formed (14, 29). The peak Ca2+ rise following the addition of sperm was $1705 \pm 119 \text{ nM Ca}^{2+}$ (SEM, n=11). Addition of 100 μ M 7-deaza-8-bromocADPR in the bathing medium (5 minutes prior to sperm addition), completely abolished the fertilisation-induced Ca2+ response (filled circles). However, the same concentration, 100 μM, of 8-bromo-cADPR only reduced the Ca2+ transient (filled triangles). The reduction of the fertilisation transient with 100 μM 8-bromo-cADPR was similar to that seen using 50 μ M 7-deaza-8-bromo-cADPR (988 \pm 81 nM Ca²⁺; SEM, n=6) supporting the greater effectiveness of the 7-deaza-8-bromo-cADPR compound. Neither antagonist mobilized Ca2+ in the eggs during the 5 minute pre-incubation period.

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(B). Dose-dependent action of 7-deaza-8-bromo-cADPR and 8-bromo-cADPR on peak fertilisation-induced Ca²⁺ rise and egg activation.

For each concentration of 7-deaza-8-bromo-cADPR (circles) and 8-bromo-cADPR (triangles), 4-9 eggs in a single dish were co-injected with heparin (250 μg/ml) and Fura (2 μM). Following incubation in artificial sea water containing 7-deaza-8-bromo-cADPR or 8-bromo-cADPR for 5 minutes sperm were added and the maximum change in intracellular free Ca²+ (primary Y-axis, filled symbols) was monitored on one of the pre-injected eggs. Following recovery of the Ca²+ transient we scored the presence or absence of an activation envelope in all injected eggs. The results (secondary Y-axis, open symbols) show that in heparinized eggs bathed in artificial sea water containing either 7-deaza-8-bromo-cADPR or 8-bromo-cADPR, there was a dose dependent reduction in the % of the eggs activated. 7-deaza-8-bromo-cADPR appeared to be more effective at all concentrations. Remaining non-heparinized eggs were all consistently activated, indicating that sperm activity was unaffected by increasing concentrations of either antagonist.

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CLAIMS

5 1. A compound having the formula (1)

where at least one of X^3 and X^7 is CR and any remaining X^3 or X^7 is N,

Y is halo, C1 - C20 hydrocarbon, NR₂, OR, SR, nitro or carboxyl,

R is H or C1 - C20 hydrocarbon and R groups can be the same or different.

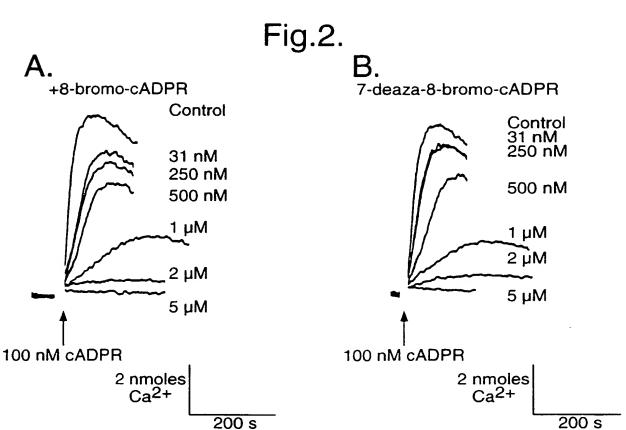
and one or each Z represents H or one Z is a cageing group.

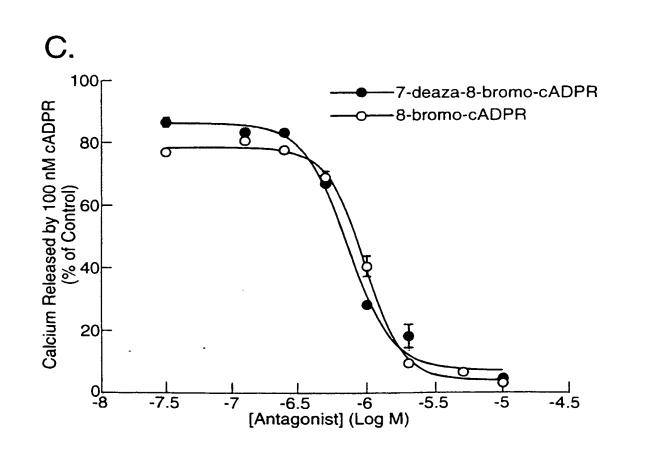
- 15 2. 7-Deaza-8-bromo-cyclic adenosine 5'-diphosphate ribose.
 - 3. A method of screening for a compound which binds to a cADPR receptor, which method comprises performing a competitive binding assay in which the compound is caused to compete with a compound according to claim 1 or claim 2 for binding to the cADPR receptor.

7-deaza-cADPR 8-bromo-cADPR 7-deaza-8-bromo-cADPR

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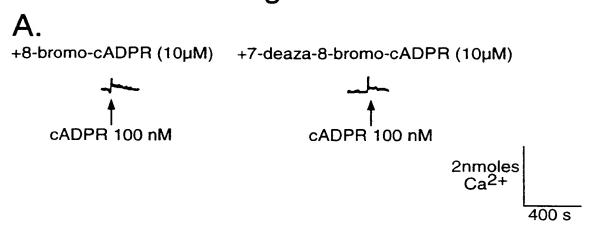
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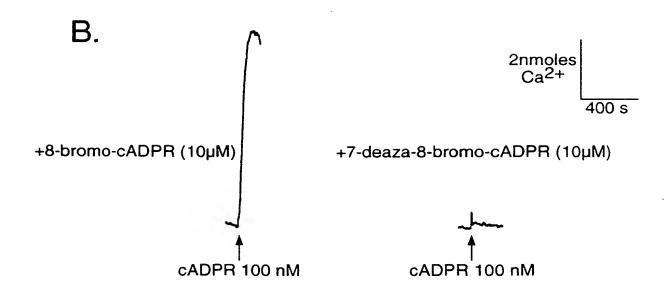




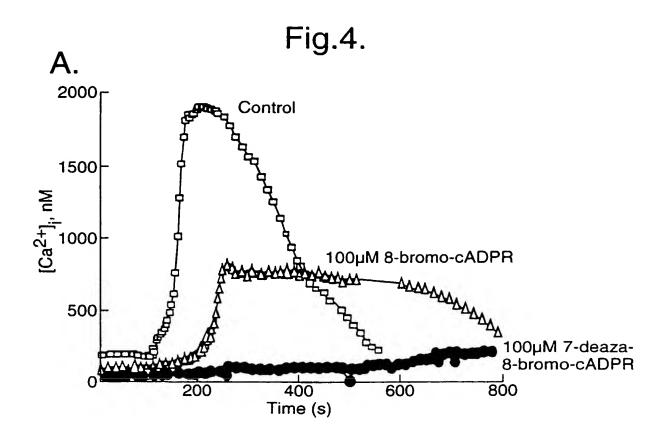
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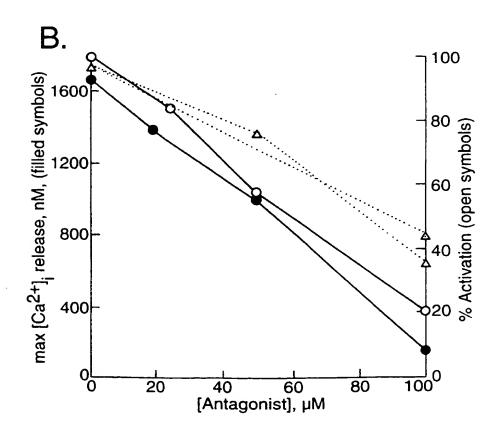
Fig.3.





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